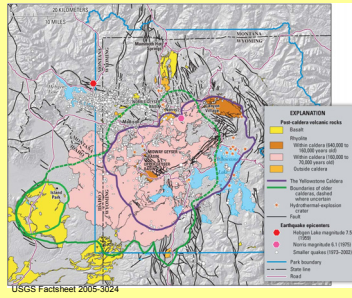
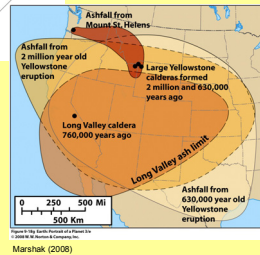
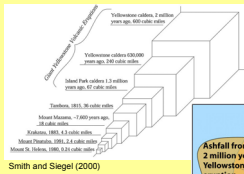
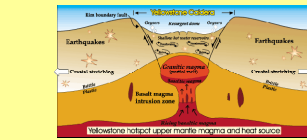
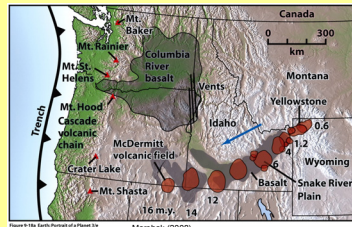


WATER CONTENTS OF YELLOWSTONE MAGMAS ESTIMATED FROM HYDROXYL CONCENTRATIONS IN FELDSPAR PHENOCRYSTS

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Background

Discovered by Ferdinand V. Hayden, Yellowstone National Park is known for its captivating scenery full of breathtaking geologic features produced by caldera-forming volcanic eruptions over the past 2 million years. Yellowstone was formed by a continental hotspot in the North American Plate about 2 million years ago after the first of three cataclysmic volcanic eruptions. Because the Yellowstone hotspot is under a continental plate, basaltic magma heats silica-rich rock creating viscous rhyolitic magma. Many explosive eruptions (for example, Mount St. Helens) occur in part because magmatic water reduces magma viscosity and forms bubbles during decompression. In this study we evaluate the hypothesis that the mega-eruptions at Yellowstone were caused by magmas with high water concentrations.



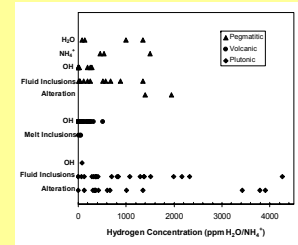
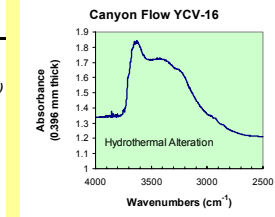
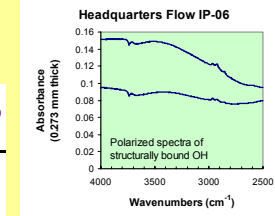
Results

- Magmatic [OH] in Yellowstone feldspars is low (12-40 ppm H₂O wt.).
- Estimated water concentration in Yellowstone magmas: less than 1.5 wt%.
- Canyon Flow feldspars: evidence for hydrothermal alteration after eruption.

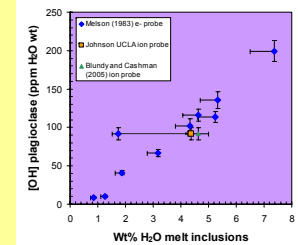
Stratigraphy, Age and OH Concentration Table				
	Unit Name*	Age*	Sample #	[OH] ppm H ₂ O by weight
First Caldera	Huckleberry Ridge Tuff	2,053 Ma		
	<i>Headquarters flow</i>	<i>1.83 Ma</i>	<i>IP-06</i>	<i>40</i>
	<i>Blue Creek flow</i>	<i>1.77 Ma</i>	<i>IP-05</i>	<i>47</i>
	Bishop Mtn flow	1.20 Ma		
	Green Canyon Pass flow	1.17 Ma		
Second Caldera	Mesa Falls Tuff	1,292 Ma		
	Osborne Butte dome	1,28 Ma		
Third Caldera	<i>Lava Creek Tuff</i>	<i>640 Ka</i>	<i>YCV-19</i>	<i>21</i>
	<i>Biscuit Basin flow</i>	<i>516 Ka</i>	<i>YCV-11</i>	<i>12</i>
	Dunraven flow	486 Ka		
	<i>Canyon flow</i>	<i>484 Ka</i>	<i>YCV-16</i>	<i>760(alteration)</i>
	Tuff of Sulphur Creek	479 Ka		
	Scaup Lake flow	198 Ka		
	Obsidian Cliff flow	183 Ka		
	Dry Creek flow	162 Ka		
	Mallard Lake flow	151 Ka		
	Summit Lake flow	112 Ka		
	Solfatera Plateau flow	110 Ka		
	West Yellowstone flow	108 Ka		
	Hayden Valley flow	102 Ka		
	Gibbon River flow	90 Ka		
Grants Pass flow	72 Ka			
Pitchstone Plateau flow	70 Ka			

*Age dates and stratigraphy from Christiansen 2001
 Bold italics indicates samples tested in this study

Infrared Spectra of Feldspars



Water species and concentrations in natural feldspars. [OH] in volcanic feldspars ranges from 0-512 ppm H₂O (Johnson 2004)

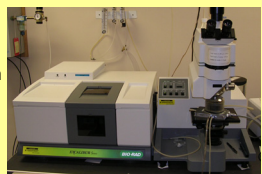


Relationship between [OH] in feldspar and H₂O in magma (Johnson 2005)

Methods

Sample Preparation

Feldspars were separated from the rock matrix by crushing, and were distinguished from quartz grains using cleavage and optical interference figures. The feldspar phenocrysts were prepared for infrared analysis by creating two perpendicular doubly-polished thick sections of each crystal. Individual phenocrysts were attached to a brass plug using Crystalbond™ epoxy. Polishing was done using aluminum oxide polishing film with grit sizes from 30 μm to 1 μm. After polishing, the samples were removed from the brass plug by heating and the remaining Crystalbond™ was removed by dissolving in acetone. Sample thickness was measured using a digital micrometer.



Infrared Analysis

Polarized infrared spectra were obtained at 4 cm⁻¹ resolution using the microscope accessory on the Varian Digilab Excalibur FTS3000 Fourier-Transform Infrared (FTIR) spectrometer in the Department of Mineral Sciences at the National Museum of Natural History, Smithsonian Institution, Washington, D.C. A modified form of the Beer-Lambert law was used to determine the concentration of OH from the IR spectra of each feldspar: $A_t = \epsilon \cdot c \cdot t$ where A_t is the total integrated IR band area in the OH region, c is the OH concentration, t is the thickness of the polished slab and ϵ' is the integrated absorption coefficient from Johnson and Rossman (2003).

Abstract

The water contents of five eruptions of Yellowstone Volcano, Wyoming, (Headquarters Flow, Blue Creek Flow, Lava Creek Tuff, Biscuit Basin Flow, and Canyon Flow) were estimated using measurements of structurally incorporated hydroxyl (OH) in feldspar phenocrysts. Hydroxyl concentrations were shown to be related to magmatic water concentrations in a previous study of feldspars from the 1980-81 eruptions of Mount St. Helens (Johnson 2005 GCA 69:A743). Feldspars from the Yellowstone samples were separated from the rock matrix by crushing and picking individual crystals, and were identified using an optical microscope. The feldspar phenocrysts were prepared for infrared analysis by creating two perpendicular doubly-polished thick sections of each crystal. Polarized infrared spectra were obtained at 4 cm⁻¹ resolution using the microscope accessory on the Varian Digilab Excalibur FTS3000 Fourier-Transform Infrared (FTIR) spectrometer in the Department of Mineral Sciences at the National Museum of Natural History, Smithsonian Institution, Washington, D.C. Hydroxyl concentrations were determined using the calibration from Johnson and Rossman (2003; Am Min v.88, 901-911). The Canyon Flow feldspars contain about 760 ppm of hydrous layer silicates and fluid inclusions indicating that these phenocrysts experienced hydrothermal exchange after eruption. The structural hydroxyl concentrations in the other four Yellowstone samples range from 12 to 47 ppm H₂O by weight. The feldspar from the Lava Creek Tuff, erupted 630,000 years ago, contains 21 ppm H₂O. These hydroxyl concentrations are on the lower end of the range of hydroxyl concentrations reported for feldspar phenocrysts from other volcanoes (0-510 ppm H₂O). Based on these measurements, we estimate that the Yellowstone magmas contained less than 1.5 wt% water.

Conclusions

- Water concentrations in the magmas were low and probably not a driving factor in the Yellowstone mega-eruptions.
- Our research contributes to the process of elimination in finding out what caused these mega-eruptions to be so enormous.
- Other techniques are needed to evaluate the role of other gases in the eruptions (Lowenstern and Hurwitz 2008).

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